

Regn No: _____

Name : _____
(To be written by the candidate)

**16th NATIONAL CERTIFICATION EXAMINATION
FOR
ENERGY MANAGERS & ENERGY AUDITORS – September, 2015**

PAPER – 2: Energy Efficiency in Thermal Utilities

Date: 19.09.2015 Timings: 1400-1700 HRS Duration: 3 HRS Max. Marks: 150

General instructions:

- Please check that this question paper contains 8 printed pages
- Please check that this question paper contains **64** questions
- The question paper is divided into three sections
- All questions in all three sections are compulsory
- All parts of a question should be answered at one place

Section – I: OBJECTIVE TYPE**Marks: 50 x 1 = 50**

- a) Answer all **50** questions
- b) Each question carries **one** mark
- c) Please hatch the appropriate oval in the OMR answer sheet with Black Pen or HB pencil

1.	Which one of the following is not true with respect to the role of nitrogen in the combustion of any fuel a) produces oxides of nitrogen b) <u>reduces the volume of combustion by-products</u> c) dilutes the flue gases d) carries useful heat in flue gases
2.	Which one of the following fuel has the highest hydrogen content and lowest sulphur content a) coal b) fuel oil c) <u>natural gas</u> d) LSHS
3.	Which of the following requires the largest amount of oxygen/kg of substance for combustion a) carbon b) <u>hydrogen</u> c) sulphur d) nitrogen
4.	Which of the following depends on physical properties of fluids as well as geometry of the heat exchanger a) <u>overall heat transfer coefficient</u> b) fouling coefficient

	c) LMTD (Log Mean Temperature Difference) d) effectiveness
5.	Which of the following can be used as desiccant in boiler preservation a) silica gel b) activated carbon c) un-slaked lime d) <u>all of the above</u>
6.	Which of the following boiler utilizes the combination of suspension firing and grate firing a) traveling grate stoker boiler b) packaged boiler c) <u>spreader stoker boiler</u> d) pulverized fuel boiler
7.	Which of the component is common to supercritical boiler and sub critical boiler for power generation a) economizer b) water walls c) re-heaters d) <u>all of the above</u>
8.	Transfer of heat without a conveying medium is possible with a) conduction b) <u>radiation</u> c) convection d) none of the above
9.	The velocity of steam in steam pipe is directly proportional to a) number of bends in pipe b) 5 th power of the diameter of pipe c) length of pipe d) <u>specific volume of steam</u>
10.	The unit of overall heat transfer coefficient is a) <u>W/m² K</u> b) W ² /m ² K c) W ² /m ³ K d) W/m ³ K
11.	The thermal resistance of an insulation a) <u>decreases with increased thermal conductivity</u> b) increases with increased thermal conductivity c) decreases with decreased thermal conductivity d) has no relation with thermal conductivity
12.	The parameter assumed to remain constant during LMTD calculation of a Heat exchanger is a) temperature drop b) heat transfer area c) <u>specific heat of fluids</u> d) none of the above
13.	The material used to control SO _x in the FBC boiler is a) <u>limestone</u> b) alumina c) silica d) fly ash
14.	Scale losses in reheating furnaces will a) <u>increase with excess air</u> b) decrease with excess air c) have no relation with excess air d) increase with CO in combustion gases
15.	NO _x formation in FBC boilers is minimised because of a) higher velocity of flue gas in combustion chamber b) higher pressure of the air supplied c) <u>lower temperatures in the bed and combustion chamber</u> d) higher contact of solid particles in the flue gas
16.	Hydrometer is used for the measurement of a) viscosity b) <u>density</u> c) water content d) humidity

17.	Heat transfer rate for indirect heating application will be less if we heat with a) saturated steam c) <u>superheated steam</u> b) dry steam d) high pressure steam
18.	Furnace wall heat loss depends on a) temperatures of external wall surfaces c) thermal conductivity of wall brick b) velocity of air around the furnace d) <u>all of the above</u>
19.	For flash steam calculation, flash steam quantity available depends upon a) <u>condensate pressure and flash steam pressure</u> c) steam enthalpy at atmospheric pressure b) steam pressure d) total heat of flash steam
20.	Condensate at pressure of 4 kg/cm ² and 160°C temperature when exposed to atmosphere will a) become super heated c) remain as condensate b) <u>partly convert to flash steam</u> d) fully convert to flash steam
21.	Coal size of 75% below 75 micron is required for use in a) spreader stoker boiler c) fluidized bed boiler b) chain grate stoker boiler d) <u>pulverized fuel boiler</u>
22.	Ceramic coating is used in furnaces because it enhances a) conductivity c) <u>emissivity</u> b) convective heat transfer coefficient d) radiation factor
23.	As the pressure of water increases from 1kg/cm ² to 8 kg/cm ² , the values of enthalpy of steam and enthalpy of evaporation respectively a) increases & remains the same c) decreases & increases b) <u>increases & decreases</u> d) decreases & remains the same
24.	A small quantity of leakage of stored Liquid LPG evaporates to produce about _____ times of volume of gas. a) 100 b) 150 c) <u>250</u> d) 350
25.	Which of the following yields a low CO ₂ reading a) soot deposits on boiler tubes c) proper air-fuel mixture b) flue gas bypass in air pre-heater d) <u>air ingress in flue gas duct</u>
26.	Which of the following salt causes temporary hardness in water a) calcium sulphate c) calcium chloride b) <u>calcium bicarbonate</u> d) calcium nitrate
27.	Which of the following is not required for determining economic thickness of steam line a) cost of fuel b) boiler efficiency c) <u>enthalpy of steam</u> d) calorific value of fuel
28.	Which of the following has the lowest stoichiometric oxygen demand (kg/kg of fuel) a) hydrogen b) carbon c) <u>sulphur</u> d) methane
29.	Which of the following contribute to spontaneous combustion of coal

	a) low Volatile matter b) low fixed carbon c) less ash d) <u>none of the above</u>
30.	Which of the following boiler water treatment ensures complete removal of salts a) <u>demineralization</u> b) softening c) de-aeration d) all of the above
31.	Which fuel among the following needs temperature control during storage a) coal b) <u>furnace oil</u> c) diesel oil d) kerosene
32.	What happens when the float in a float trap develops a puncture a) loss of condensate b) loss of live steam c) <u>fails to open</u> d) fails to close
33.	The stoichiometric amount of air required to burn 1 kg of methane is a) 69.57 b) 4 c) <u>17.39</u> d) 16
34.	The heat loss in a furnace depends on a) emissivity of walls b) conductivity of refractory c) wall thickness d) <u>all of the above</u>
35.	The chemical used to prevent oxidation in boiler feedwater is a) <u>sodium sulphite</u> b) sodium phosphate c) calcium phosphate d) magnesium phosphate
36.	The best time for intermittent blow down in a boiler a) High load under full pressure b) <u>low load under full pressure</u> c) high load under partial pressure d) low load under partial pressure
37.	The amount of CO ₂ produced in complete combustion of 18 Kg of carbon a) 50 b) 44 c) <u>66</u> d) 792
38.	Removal of condensate from main steam line is done to prevent a) steam locking b) air locking c) <u>water hammer</u> d) all of the above
39.	Presence of _____ in flue gas confirms incomplete combustion in furnace a) <u>CO</u> b) NO _x c) SO _x d) all of the above
40.	Portable fyrite is used for the measurement of a) CO ₂ content in flue gas b) O ₂ content in flue gas c) stack temperature d) <u>both (a) & (b)</u>
41.	On an inverted bucket trap, what happens to the bucket as the trap fills with water a) it rises b) it inclines c) <u>it sinks</u> d) it remains stationary
42.	In which zone of cupola furnace does the conversion of CO ₂ to CO take place? a) combustion zone b) melting zone c) <u>reduction zone</u> d) preheating zone
43.	In a fire-tube boiler, soot forms on a) outside tube surface b) <u>inside tube surface</u> c) waterside surface d) water wall surface

44.	If the volatile matter in coal is low, which of the following equipment is the best waste heat recovery option in a boiler a) economiser b) <u>air preheater</u> c) deaerator d) heat pipe
45.	Enthalpy of evaporation of any vapour at its critical point will be a) more than zero b) <u>zero</u> c) less than zero d) unpredictable
46.	Corrosion in chimney, air pre-heater and economizer is mainly influenced by a) <u>sulphur content in fuel</u> b) ash content in fuel c) moisture content in fuel d) all of the above
47.	An increase in bulk density of a refractory increases its a) volume stability b) heat capacity c) Resistance to slag penetration d) <u>all of the above</u>
48.	A high carbon monoxide reading indicates a) moisture in the fuel b) high furnace temperature c) high excess air d) <u>none of the above</u>
49.	2 m lift of condensate in steam pipe will result in back pressure of a) 0.02 bar b) <u>0.2 bar</u> c) 2 bar d) 20 bar
50.	_____ gives an estimate of heating value of coal a) ash content b) moisture content c) <u>fixed carbon</u> d) volatile matter

----- End of Section - I -----

Section – II: SHORT DESCRIPTIVE QUESTIONS

Marks: 8 x 5 = 40

- (i) Answer all **Eight** questions
(ii) Each question carries **Five** marks

S-1	Calculate the electricity consumption in an induction melting furnace from the following melt cycle data, Mild steel (MS) scrap charged : 1250 kg Specific heat of MS : 0.68 kJ/kg °C Latent heat of MS : 270 kJ/kg MS melting temperature : 1650 °C Inlet MS charge temperature : 35 °C Efficiency of furnace : 65%
Ans:	Theoretical energy required for melting = $1250 (0.68 \times (1650 - 35) + 270)/3600$ = 475.1 kWh (3 marks)

	<p>Actual energy input to the furnace = 475.1 / 0.65</p> <p>Electricity consumption = 730.9 kWh</p> <p>.... (2 marks)</p>
S-2	<p>(a) Why should LPG cylinders not be stored in basements or cellars? (2.5 marks)</p> <p>(b) Why should the stack temperature of furnace oil fired boilers not be maintained below 160-170°C ? (2.5 marks)</p>
	<p>Ans:</p> <p>(a) LPG is a predominant mixture of propane and butane. Both propane and butane are denser than air. Consequently, the vapour flows along the ground into drains and sinks to the lowest level of the surroundings and gets ignited at a considerable distance from the source of leakage. Escape of even small quantities of LPG can give rise to large volume of vapour mixture and can cause considerable hazard. Hence there should be adequate ground level ventilation where LPG cylinders are stored. For this reason LPG cylinders should not be stored in cellars or basements, which have no ventilation at ground levels.</p> <p>....(2.5 marks)</p> <p>(b) For fuels containing sulphur, low temperatures (below 160-170°C) of stack can lead to sulphur dew point corrosion. The main disadvantage of sulphur is the risk of corrosion by sulphuric acid formed during and after combustion, and condensing in cool parts of the chimney or stack, air preheater and economiser.</p> <p>....(2.5 marks)</p>
S-3	<p>Feed water is provided to a boiler from the feed water tank at 60°C, temperature of condensate water returning to the tank is 88°C, and temperature of makeup water is 27°C. What is the amount of condensate recovered ?</p>
	<p>Solution:</p> <div style="text-align: center;"> </div> <p> $27x + (1 - x) 88 = 60$ Therefore $x = 0.36$ (36 percent makeup water or only 64% of condensate is recovered). </p> <p>....(5 marks)</p>

<p>S-4</p>	<p>Milk is flowing in a pipe cooler at a rate of 0.85 kg/sec. Initial temperature of the milk is 60°C and it is cooled to 22 °C using a stirred water bath with a constant temperature of 10°C around the pipe. Specific heat of milk is 3.86 KJ/kg°C. Calculate the heat transfer rate (kcal/hr) and also LMTD of the exchanger.</p>																
	<p>Ans:</p> <p>Heat transfer in cooling milk = $0.85 * 3.86 * (60 - 22)$ $= 124.678$ KJ/sec $= 124.678 / 4.18$ $= 29.83$ Kcal /sec $= (29.83*3600)= 107378.122$ kCal/hr <p style="text-align: right;">.... (2.5 marks)</p> <p>LMTD:</p> <p>DT1 = 60 – 10 = 50 °C</p> <p>DT2 = 22 – 10 = 12 °C</p> <p>LMT D of the heat exchanger = $(50 - 12) / \ln (50 / 12)$ $= 26.6$ °C <p style="text-align: right;">.... (2.5 marks)</p> </p></p>																
<p>S-5</p>	<p>Paddy husk is being used as a combustion fuel in a water tube boiler. The ultimate analysis of fuel is given below. Calculate theoretical amount of air required per 100 kg of husk for the combustion from the following data.</p> <table border="1" data-bbox="427 1451 1114 1798"> <thead> <tr> <th>Ultimate Analysis of Typical Agro Residues</th> <th>%</th> </tr> </thead> <tbody> <tr> <td>Moisture</td> <td>11.8</td> </tr> <tr> <td>Mineral Matter</td> <td>17.7</td> </tr> <tr> <td>Carbon</td> <td>32.0</td> </tr> <tr> <td>Hydrogen</td> <td>5.0</td> </tr> <tr> <td>Nitrogen</td> <td>0.9</td> </tr> <tr> <td>Sulphur</td> <td>0.1</td> </tr> <tr> <td>Oxygen</td> <td>32.5</td> </tr> </tbody> </table>	Ultimate Analysis of Typical Agro Residues	%	Moisture	11.8	Mineral Matter	17.7	Carbon	32.0	Hydrogen	5.0	Nitrogen	0.9	Sulphur	0.1	Oxygen	32.5
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	<p>Ans</p> <p>Considering a sample of 100 kg of paddy husk.</p>																

Component in fuel	% (wt) in fuel	Equation	Specific stoichiometric oxygen required	Actual stoichiometric oxygen required
Carbon	32	$C + O_2 = CO_2$ $12 + 32 = 44$	2.67 (32/12)	85.44 (32x2.67)
Hydrogen	5	$H_2 + 0.5 O_2 = H_2O$ $2 + 16 = 18$	8.0 (16/2)	40.0 (5x8)
Sulphur	0.1	$S + O_2 = SO_2$ $32 + 32 = 64$	1.0 (32/32)	0.1 (0.1x1)
Oxygen	32.5			(-)32.5
Nitrogen	0.9			-
Moisture	11.8			-
Mineral matter	17.7			-
Total	100			93.04

.... (4 marks)

Total Oxygen required = 93.04 kg oxygen / 100 kg fuel

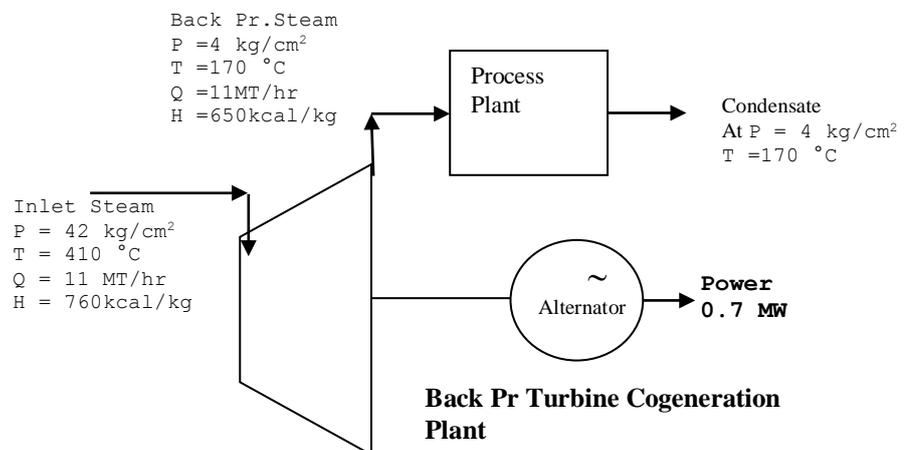
Therefore theoretical quantity of dry air reqd. = 93.04 / 0.23 = 404.5 kg air / 100 kg fuel

(air contains 23% oxygen by wt.)

.....(1 mark)

S-6

Determine the Energy Utilization Factor (EUF) from the following back pressure cogeneration plant diagram and data given.



	<p>Solution</p> <p>Input heat to turbine = 11,000 x 760 = 8360000 Kcal/hr(1 mark)</p> <p>Useful heat to process Plant = 11,000 x 650-170 = 5280000 Kcal/hr(1 mark)</p> <p>Useful Electrical output in alternator = 700x860= 602000 Kcal/hr(1 mark)</p> <p>Energy Utilization Factor (EUF) = (602000+5280000) /8360000 = 70.36%(2 marks)</p>
S-7	<p>For combustion of 500 kg/hr of Natural Gas containing 100% Methane, Calculate the percentage of CO₂ in the flue gas while 20% excess air is supplied.</p>
	<p>Ans:</p> <p>$\text{CH}_4 + 2 \text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$</p> <p>1 mole of Methane requires 2 moles of Oxygen.</p> <p>16 Kg of Methane requires 64 Kg of Oxygen.</p> <p>16 Kg of Methane produces 44 Kg of CO₂.</p> <p>500 Kg/hr of Methane requires 2000 Kg/hr of Oxygen.</p> <p>500 Kg/hr of Methane requires 1375 Kg/hr of CO₂. (1 mark)</p> <p>Theoretical air required for combustion = 2000 / 0.23 = 8695 Kg/hr (1 mark)</p> <p>Considering 20% excess air,</p> <p>Actual air supplied for combustion is = 8696 * 1.2 = 10435 Kg/hr of air (1 mark)</p> <p>Flue gas generation with 20% excess air = 500 + 10435 = 10935 Kg/hr (1 mark)</p> <p>% CO₂ in the flue gas = 1375 / 10935 = 12.6 % (1 mark)</p>
S-8	<p>In a sugar mill, a process requires 5000 kg/hr of dry saturated steam at 7 kg/cm² (g). For the flow velocity not to exceed 25 m/s, determine the pipe diameter size for distribution of steam.</p>

	Specific volume at 7 kg/cm ² = 0.24 m ³ /kg
	Ans.
	The velocity of steam maximum = 25 m/s
	Specific volume at 7 kg/cm ² = 0.24 m ³ /kg
	Mass flow rate = 5000 kg/hr
	= 5000/3600 = 1.389 kg/sec
	Volumetric flow = 1.389 x 0.24 = 0.333 m ³ /sec
 (1 mark)
	Volumetric flow rate(m ³ /s) = Velocity (m/s) x cross sectional area (m ²)
 (1 mark)
	Therefore,
	$D = \sqrt{\frac{4 \times \text{Volumetric flowrate}}{\pi \times \text{Flow velocity}}}$
	$D = \sqrt{\frac{4 \times 0.333}{\pi \times 25}}$
	$D = 0.130 \text{ m or } 130 \text{ mm}$
 (3 marks)
	Since the steam velocity must not exceed 25 m/s, the pipe size must be at least 130 mm; the nearest commercially available size, 150 mm, would be selected.

----- End of Section - II -----

Section - III: LONG DESCRIPTIVE QUESTIONS

Marks: 6 x 10 = 60

- (i) Answer all SIX questions
- (ii) Each question carries **Ten** marks

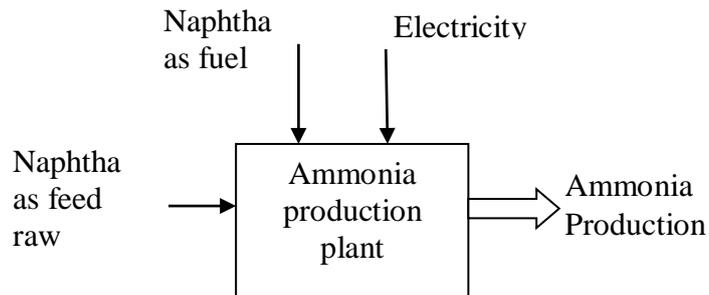
L-1	Write short notes on any two of the following: (5 marks each)
	a) Plate heat exchanger (page 242 of book-2)
	b) Multiple effect evaporator (page 247-248 of book-2)
	c) Gas turbine cogeneration system (page 192 of book-2)

L-2	<p>a) Find out the efficiency of the furnace oil fired boiler by the direct method in an agro product manufacturing plant given the following data:</p> <table style="width: 100%; border: none;"> <tr><td>Type of boiler</td><td>: Furnace oil fired</td></tr> <tr><td>Quantity of steam (dry) generated</td><td>: 5 TPH</td></tr> <tr><td>Steam pressure / temp</td><td>: 10 kg/cm²(g)/ 180 °C</td></tr> <tr><td>Quantity of oil consumed</td><td>: 0.350 TPH</td></tr> <tr><td>Feed water temperature</td><td>: 75 °C</td></tr> <tr><td>GCV of Furnace oil</td><td>: 10300 kCal/kg</td></tr> <tr><td>Enthalpy of saturated steam at 10 kg/cm² pressure</td><td>: 665 kCal/kg</td></tr> <tr><td>Enthalpy of feed water</td><td>: 75 kCal/kg</td></tr> <tr><td>Cost of furnace oil</td><td>: Rs 32/kg</td></tr> <tr><td>Annual operating hours</td><td>: 7200 hrs /year</td></tr> </table> <p>b) The oil fired boiler was converted to coconut shell firing maintaining the same steam and feed water parameters.</p> <p>i) Determine the fuel consumption per hour ii) Return on investment for the conversion scheme.</p> <table style="width: 100%; border: none;"> <tr><td>Fuel fired in the boiler</td><td>: coconut shell fuel</td></tr> <tr><td>GCV of coconut shell</td><td>: 4200 kCal/kg</td></tr> <tr><td>Efficiency with coconut shell firing</td><td>: 76%</td></tr> <tr><td>Cost of coconut shell</td><td>: Rs 12/kg</td></tr> <tr><td>Annual interest on capital</td><td>: Rs 6 lakhs /yr</td></tr> <tr><td>Annual operating hours</td><td>: 7200 hrs /year</td></tr> <tr><td>Investment towards boiler conversion</td><td>: Rs 50 lakhs</td></tr> </table>	Type of boiler	: Furnace oil fired	Quantity of steam (dry) generated	: 5 TPH	Steam pressure / temp	: 10 kg/cm ² (g)/ 180 °C	Quantity of oil consumed	: 0.350 TPH	Feed water temperature	: 75 °C	GCV of Furnace oil	: 10300 kCal/kg	Enthalpy of saturated steam at 10 kg/cm ² pressure	: 665 kCal/kg	Enthalpy of feed water	: 75 kCal/kg	Cost of furnace oil	: Rs 32/kg	Annual operating hours	: 7200 hrs /year	Fuel fired in the boiler	: coconut shell fuel	GCV of coconut shell	: 4200 kCal/kg	Efficiency with coconut shell firing	: 76%	Cost of coconut shell	: Rs 12/kg	Annual interest on capital	: Rs 6 lakhs /yr	Annual operating hours	: 7200 hrs /year	Investment towards boiler conversion	: Rs 50 lakhs
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	<p>Solution:</p> <p>a) Efficiency of furnace oil fired boiler (Direct method)</p> $\text{Boiler Efficiency } (\eta) = \frac{5000 \times (665-75)}{350 \times 10300} = 81.8\% \text{ (on GCV basis)}$ <p style="text-align: right;">.... (2.5 marks)</p> <p>b) i) Coconut shell fuel consumption after conversion:</p> $\text{Fuel consumption} = \frac{5000 \times (665-75)}{0.76 \times 4200} = 924.2 \text{ kg/hr}$ <p style="text-align: right;">.... (2.5 marks)</p> <p>ii) ROI for the conversion scheme:</p> <p>Annual fuel cost of furnace oil fired boiler = $350 \times 7200 \times 32 = \text{Rs } 8,06,40,000$ /year (1 mark)</p> <p>Annual fuel cost of coconut shell fired boiler = $924.2 \times 7200 \times 12 = \text{Rs } 7,98,50,880$/year (1 mark)</p> <p>Annual net monetary savings after conversion</p> $= \frac{[(8,06,40,000 - 7,98,50,880) - 6,00,000]}{50,00,000} \times 100 = 3.8\%$																																		

..... (3 marks)

L-3

a) In a typical fertiliser manufacturing plant, the quantity of 133200 Ton of Ammonia is produced using naphtha as fuel as well as raw material (feed) and electricity from captive power plant.



The quantity of raw material consumption and its heating values are given in table.

Raw material consumption	Quantity	Heating Value
1.Naphtha - feed	56800 Ton	10650 kCal / kg
2.Naphtha - fuel	28200 Ton	10650 kCal / kg
3.Electricity	1080 x Lakh kWh	3200 kCal/kWh

Calculate the specific energy consumption of ammonia production in Gcal / Ton.

b) Sketch the schematic diagram of “Back Pressure Turbine” and “Extraction Condensing Turbine” Cogeneration systems (Note: no explanation required).

Answer

a) **Specific Energy Consumption(SEC) of Ammonia**

Raw material consumption	Quantity	Heating Value	Gcal
1.Naphtha - feed	56800 Ton	10650 kCal / kg	604920
2.Naphtha - fuel	28200 Ton	10650 kCal / kg	300330
3.Electricity	1080 x Lakh	3200 kCal/kWh	345600

1 mark

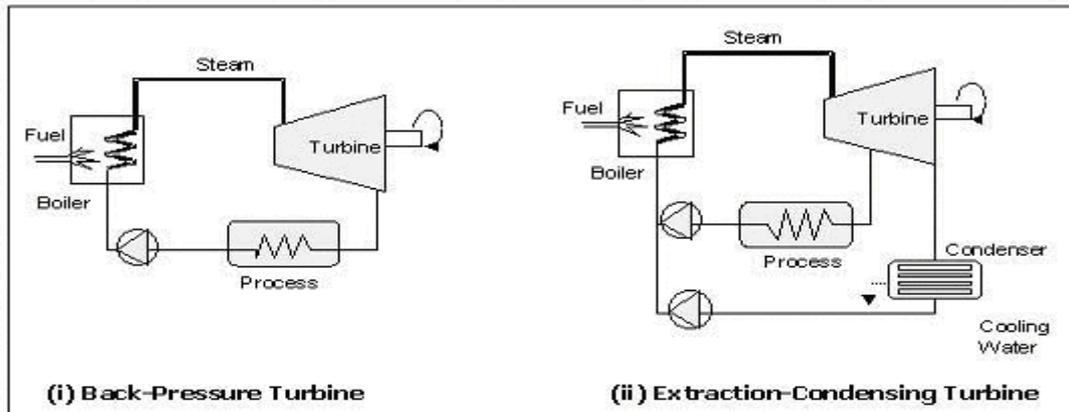
1 mark

1 mark

	kWh			
				1250850
Ammonia production	133200 Ton			
Specific Energy Consumption(SEC) of Ammonia production =			9.39 Gcal per Ton	2 marks

Answer

b) schematic diagram of “(i)Back Pressure Turbine” and “(ii)Extraction Condensing Turbine” Cogeneration systems

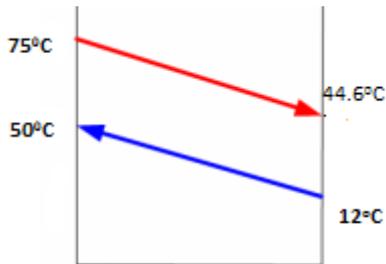


.....(2.5 marks for each schematic)

L-4	<p>As a part of energy conservation measure, APH (Air Pre-heater) is installed in a fired heater. The APH is designed to pre-heat 240 m³/min of combustion air to 250°C. Flue gas enters the APH at 375°C. Calculate the flue gas leaving the stack and also determine the improvement in furnace efficiency after the installation of APH with the following data</p> <p>Density of air : 1.15 kg/m³</p> <p>Specific heat of air : 0.23 Kcal/kg°C</p> <p>Specific heat of flue gas : 0.26 Kcal/kg°C</p> <p>Calorific value of fuel : 9850 Kcal/kg</p> <p>Air to fuel ratio : 17</p> <p>Efficiency of furnace : 69 %</p> <p>Ambient temperature : 30°C</p>
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	<p>Solution:</p> <p>Amount of Air flow = $240 * 60 * 1.15$ = 16560 Kg/hr (1mark)</p> <p>Amount of fuel = $16560 / 17$ = 974.12 Kg/hr (1mark)</p> <p>Amount of flue gas = $16560 + 974.12$ = 17534.1 Kg/hr (1mark)</p> <p>Heat absorbed by combustion air = $16560 * 0.23 * (250 - 30)$ = 837936 Kcal/hr (1.5 mark)</p> <p>Temperature difference in flue gas = $837936 / (17534.1 * 0.26)$ = 183.8 °C (1.5 mark)</p> <p>Flue gas leaves the stack at temp = $375 - 183.8 = 191.2$ °C (1 mark)</p> <p>Efficiency of APH = Heat absorbed by air / Heat input * 100 = $837936 * 100 / (974.12 * 9850)$ = 8.73 % (2 marks)</p> <p>Overall efficiency after APH = $69 + 8.73 \% = 77.73 \%$ (1 mark)</p>
L-5	<p>A liquid waste stream has a flow rate of 3.5 kg/s and a temperature of 75° C with a specific heat capacity of 4190 J/kgK. Heat recovered from the hot waste stream is used to pre-heat boiler make-up water. The flow rate of the make-up water is 2.8 kg/s, its temperature is 12°C and its specific heat capacity is 4190 J/kg/K. The overall heat transfer coefficient of the heat exchanger is 800 W/m²K. If a make-up water exit temperature of 50°C is required, and assuming that there are no heat loss from the heat losses from the exchanger, determine</p> <ol style="list-style-type: none"> 1) The heat transfer rate 2) The exit temperature of the effluent and 3) The area of the heat exchanger required
	<p>Solution :</p> <p>i) Heat gained by makeup water = $Q_c = m_c c_c \Delta T = 2.8 * 4190 * (50-12)$ = 445816 W = 445.82 kW (2 marks)</p> <p>ii) $m_h C_h (t_{h1} - t_{h2}) = m_c C_c (t_{c1} - t_{c2})$ $3.5 * 4190 * (75 - t_{h2}) = 2.8 * 4190 * (50 - 12)$ $t_{h2} = 44.6$ ° C(3 marks)</p>

iii) Now because the water outlet temperature is above the outlet temperature of the effluent a counter-flow heat exchanger is required



Counter Flow Arrangement

$$\begin{aligned}
 \text{LMTD} &= \frac{\Delta t_1 - \Delta t_2}{\ln (\Delta t_1 / \Delta t_2)} \\
 &= \frac{(75-50)-(44.6-12)}{\ln (75-50)/(44.6-12)} \\
 &= \frac{25-32.4}{\ln(25/32.4)}
 \end{aligned}$$

.....(2.5 marks)

$$\text{LMTD} = 28.54^\circ\text{C}$$

$$\begin{aligned}
 Q &= UA (\text{LMTD}) \\
 &= \frac{445816}{800 \times 28.54}
 \end{aligned}$$

$$\text{Area} = 19.5 \text{ m}^2$$

.....(2.5 marks)

L-6

A process industry is equipped with a steam power plant generating 1,25,000 units/day and a separate low-pressure boiler generating at an average steam production of 9.5 Tons of steam/hour at enthalpy of 630 kcal/kg for process heating. The feed water temperature to the boiler is 62°C. The efficiencies of the steam power plant and boiler are 27% and 78% respectively. Coal is used in both cases and calorific value of coal is 4000 kcal/kg.

The management proposes to commission a cogeneration plant retaining the coal as fuel. The expected energy utilization factor of the cogeneration plant is 75%.

Calculate coal savings with the new cogeneration plant.

SOLUTION:

Heat input for separate production of power and useful heat	=	$(125000 \times 860) / (24 \times 0.27) + (9.5 \times (630-62) \times 1000) / 0.78$
	=	16589506 + 6917949
	=	23507455 kcal/hr
	 (4 marks)

	Heat input with cogeneration plant	=	$[(125000 \times 860) / 24 + (9.5 \times (630-62) \times 1000)] / 0.75$
		=	$(4479167 + 5396000) / 0.75$
		=	13166889 kcal/hr
		(4 marks)
	Coal savings	=	$(23507455 - 13166889) / 4000$
		=	2585 kg/h
		(2 marks)

..... **End of Section – III**